## **Final Report**

## PD-111900

### Dynamics of polyatomic chemical reactions

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# Introduction

The research supported by the present postdoctoral grant started in January of 2015 when I was a research associate at the Institute of Chemistry, Eötvös University. Then I was invited to join the faculty of the University of Szeged and from August of 2015 I continued the present project at the Department of Physical Chemistry and Materials Science of the University of Szeged. Using the support of a postdoctoral grant I established a new Computational Reaction Dynamics Research Group in Szeged. We published 15 papers including two *Nature Chemistry* articles and I gave several invited talks at international conferences with the support of the PD-111900 grant. Furthermore, at the end of 2015 I submitted my Doctor of the Hungarian Academy of Sciences dissertation, which was successfully defended in January, 2017. In 2017 I received a K NKFIH grant based partially on my achievements during the present PD-grant period.

The main goal of our research is to study the dynamics and mechanisms of polyatomic chemical reactions using ab initio potential energy surfaces (PESs) and the quasi-classical trajectory (QCT) method. We aim to reveal new reaction pathways as well as understand the energy efficiency and mode-specificity of chemical systems, which may help to control the outcome of chemical reactions. The summary of our research achievements is described below in chronological order.

#### Results

**Double-inversion:** A new reaction pathway for  $S_N 2$  reactions. In January 2015 we published a reaction dynamics study revealing a new retention mechanism, called double



Figure 1. The double-inversion mechanism of  $S_N 2$  reactions.

inversion, for the  $F^-$  + CH<sub>3</sub>Cl S<sub>N</sub>2 reaction [*Nat. Commun.* **6**, 5972 (2015)]. During double inversion a proton-abstractioninduced inversion is followed by a second inversion via the usual central transition state, thereby resulting in retention of configuration (Figure 1). Our finding received significant media attention; for example, the National Geographic Hungary, Index.hu, and the Hungarian Television highlighted the new reaction path. Even if the *Nature Communications* article on the new double-inversion mechanism appeared in 2015, this paper is not listed among the PD-111900-supported publications, because the work had been performed before the grant opened. Nevertheless, it is important to mention this double-inversion study, because the new reaction pathway provided motivation for other projects reviewed in this report. For example, in 2015 we investigated the generality of the new double-inversion mechanism [1] and found the so-called double-inversion transition state for all the 16  $S_N 2$  reactions of  $X^- + CH_3 Y$  type, where X and Y = F, Cl, Br, I. We claimed that double inversion may be a general mechanism for  $S_N 2$  reactions.

**Spin-orbit effects for a chemical reaction.** Beside the above-discussed  $S_N2$  reactions, I had collaboration with the group of Professor Schaefer (University of Georgia, GA, USA). We investigated the pathways for the OH + Cl<sub>2</sub>  $\rightarrow$  HOCl + Cl and HOCl + Cl  $\rightarrow$  HCl + ClO reactions [2]. Our benchmark study considering high-level ab initio computations including spin-orbit corrections provided the best theoretical estimates and predictions of the energetics of the OH + Cl<sub>2</sub> system.

A global PES and dynamics for  $F^- + CH_3F$ . We developed a global, fulldimensional, chemically accurate, ab initio PES for the  $F^- + CH_3F$  system, which PES describes the Walden-inversion, front-side attack, and double-inversion  $S_N2$  pathways as well as the proton-abstraction channel (HF + CH<sub>2</sub>F<sup>-</sup>) [3]. Quasi-classical trajectory simulations on the new PES showed that front-side attack is a significant mechanism at high collision energies. Double inversion can also occur in the  $F^- + CH_3F$  reaction, though the doubleinversion cross sections are small. Furthermore, the simulations found significant amount (around 20 %) of re-crossing trajectories showing that traditional transition-state-theory computations may overestimate the reactivity.

**Rotational mode specificity in S<sub>N</sub>2 reactions.** We investigated, for the first time, the rotational mode specificity (*JK*-dependent reactivity) in S<sub>N</sub>2 reactions [4]. For the  $F^- + CH_3Y(v = 0, JK)$  [Y = F and Cl] S<sub>N</sub>2 reactions we found that reactant rotational excitations usually decrease the reactivity, because the rotational motion counteracts the ion-dipole orientation effects. Further studies, inspired by ours, would be necessary to develop general qualitative rules for rotational mode specificity.

**Unexpected leaving-group effect in S<sub>N</sub>2 reactions.** In 2016 we published a joint experimental-theoretical work in *Nature Chemistry* on the leaving-group effects on the dynamics of S<sub>N</sub>2 reactions [5]. We found that the leaving group can have different influence on the dynamics of seemingly similar S<sub>N</sub>2 reactions, e.g.,  $F^-$  + CH<sub>3</sub>Cl and  $F^-$  + CH<sub>3</sub>I. Furthermore, our theoretical results were in unprecedented agreement with the experimental data for a S<sub>N</sub>2 reaction. These findings were highlighted by the national (delmagyar.hu) and international (phys.org) media.

The best quantum dynamics for a S<sub>N</sub>2 reaction. Utilizing our full-dimensional analytical PES we studied the mode-specific dynamics of the  $F^- + CH_3Cl S_N2$ reaction using reduced-dimensional quantum as well as full-dimensional QCT methods. The present sixdimensional quantum simulations, carried out in collaboration with the group of Professor Minghui Yang, represent the highest-dimensional quantum dynamical study of a  $S_N 2$  reaction. Despite the fact that the above  $S_N 2$ reaction is barrierless (the Walden-inversion transition



**Figure 2.** Quantum *vs.* classical dynamics of  $F^-$  + CH<sub>3</sub>Cl.

state is energetically below the reactants), the quantum computations revealed significant mode-specificity, in good agreement with the QCT results. The quantum-mechanical and QCT excitation functions (integral cross sections as a function of collision energy) obtained for the ground-state  $F^- + CH_3Cl(v = 0) S_N2$  reaction were found to agree very well supporting the validity of the QCT method for  $S_N2$  reactions (Figure 2). The results were published in the high-impact physical chemistry journal *J. Phys. Chem. Lett.* [6].

Mode-specific dynamics of  $F^-$  + CHD<sub>2</sub>Cl. In 2016 we also investigated the multichannel mode-specific dynamics of the  $F^-$  + CHD<sub>2</sub>Cl S<sub>N</sub>2 and proton/deuteron-transfer reactions using our full-dimensional global ab initio PES [7]. QCT computations on the PES show that double inversion is promoted if we excite the CH or CD stretching mode of the reactant. Considering the proton/deuteron-transfer channels leading to either HF +  $CD_2Cl^-$  or DF + CHDCl<sup>-</sup>, the proton abstraction is found more likely than deuteron abstraction and the CH and CD stretching excitations enhance the HF and DF channels, respectively. Correlated scattering- and attack-angle distributions reveal that the direct abstraction processes usually go through back-side rebound, back-side stripping, and front-side stripping pathways. The CH and CD stretching modes do not behave as pure spectators in the abstraction reactions, since in all cases mainly vibrationally ground-state products are formed. The rotational distributions of the HF and DF products are also found to be cold, thus most of the available energy transfers into the relative translational energy of the products. In this study we have also tested the one-dimensional Gaussian binning (1GB) method, which was first proposed by us in 2009 [J. Chem. Phys. 131, 244302 (2009)], for mode-specific polyatomic product analysis. The results show that the exact classical vibrational energy-based 1GB method provides physically correct final state distributions.

The first ab initio analytical PES for  $F^- + CH_3I$ . In 2017 we reported a global PES for the  $F^- + CH_3I$  reaction which PES describes the  $S_N2$  channel via the back-side attack inversion, front-side attack retention, and double-inversion pathways, as well as the protontransfer channel [8]. The new PES is obtained by fitting high-level explicitly-correlated ab initio energy points using the permutationally invariant polynomial approach. This is the first full-dimensional ab initio analytical PES for the  $F^- + CH_3I$  system, which opens the door for many dynamical simulations. We computed the cross sections and reaction probabilities of the  $S_N2$  and proton-transfer channels as well as the scattering angle distributions and the product internal energy distributions for the  $S_N2$  channel. The computed results were compared with available crossed-beam experiments showing remarkable agreement. Furthermore, additional experiments have recently been performed for this system in Professor Roland Wester's laboratory (University of Innsbruck), and we collaborate with the Wester group on the modespecific dynamics of the  $F^- + CH_3I S_N2$  reaction. The joint experimental-theoretical results will be published soon.

 $S_N 2 vs. E2$  mechanisms. We performed a high-level ab initio benchmark study on the characterization of the stationary points of the PES of the F<sup>-</sup> + CH<sub>3</sub>CH<sub>2</sub>Cl reaction [9]. In the case of this system, besides the well-known  $S_N 2$  channel, an E2 elimination process can also occur. Furthermore, we have found a double-inversion transition state, for the first time for a reaction of an ethyl-halide. Following this ab initio study, we are currently working on

developing a global PES for this system, which will open the door for collaboration with experimentalists.

**Front-side complex formation in**  $S_N2$  **reactions.** As mentioned above in 2016 we found that the leaving group may have significant effect on the dynamics of  $S_N2$  reactions. Since in 2017 we had PESs for both the  $F^- + CH_3Cl$  and  $CH_3I$  reactions at hand, we could compare the dynamics of the two  $S_N2$  reactions. The simulations revealed significant front-side complex formation for  $F^- + CH_3I$ , which steers the reactants away from the reactive orientations, thereby making the reaction more indirect, whereas the front-side complex formation is negligible for  $F^- + CH_3Cl$ , therefore this reaction is more direct [10]. Our study provides the first quantitative characterization of the front-side complex formation in the  $S_N2$  process and highlights the role of these non-traditional minima in the dynamics of  $S_N2$  reactions.

**Benchmark ab initio study on CI^- + CH\_3I.** Motivated by the novel stationary points recently revealed for  $S_N2$  reactions, we performed a high-level ab initio characterization of the PES of the  $CI^- + CH_3I$   $S_N2$  and proton-transfer reactions [11]. On the qualitative side, we reported new minima and transition states for the  $CI^- + CH_3I$  system. On the quantitative side, we published definitive structures based on CCSD(T)-F12b/aug-cc-pVQZ computations as well as accurate relative energies obtained by considering electron correlation beyond the gold standard CCSD(T) level, extrapolation to the complete basis set limits, and core correlation contributions.

**Mapping the angle-dependent barrier to reaction.** The barrier of a chemical reaction plays a central role in chemistry, which has two major properties, the height and its dependence on geometry (Figure 3). The former has been widely studied, but the experimental determination of the latter has been impossible until now. In 2017 Kopin Liu's group developed a method to measure the angle-dependent barrier for the  $Cl + CHD_3$  reaction. The experimental data were excellent agreement with my computed values. In order to perform this experiment the orientation of the



**Figure 3.** Transition-state bending potential of a chemical reaction.

reactants needs to be controlled. The resulting polarization-dependent differential cross sections were found to be in very good agreement with the aligned QCT results revealing different dynamics of end-on and side-on collisions. The experimental–theoretical results were published in a joint *Nature Chemistry* paper [12] in the issue of December, 2017.

Feature Article on  $S_N2$  reactions. In 2017 we received an invitation to submit a Feature Article for *J. Phys. Chem. A* highlighting our recent research achievements. This article reviews our novel analytical PES-based approach to  $S_N2$  reaction dynamics as well as many of the above described research findings including the double-inversion mechanism, leaving-group effect, front-side complex formation, the best quantum dynamics for  $S_N2$  reactions, and the theoretical–experimental comparisons. The paper was featured on the front cover (Figure 4) of the journal [13].



Figure 4. Cover.

**Revealing novel reaction pathways for Cl** + CH<sub>4</sub>  $\rightarrow$  H + CH<sub>3</sub>Cl. QCT simulations for Cl + CH<sub>4</sub> on a new revised PES showed that significant fraction of the substitution channel does not proceed via Walden inversion, but results in retention of configuration. The trajectories revealed two different types of retention mechanisms, named edge- and vertexattack. During edge-attack the Cl atom directly replaces a side H atom, whereas the vertexattack proceeds via a H-abstraction-induced substitution. More details can be found in Ref. [14].

 $S_N 2$  reactions at nitrogen centers. Besides the above-discussed carbon-centered  $S_N 2$  reactions, we have studied  $S_N 2$  reactions at nitrogen (N) centrums. As a first step, we have characterized the  $X^- + NH_2Y$  [X, Y = F, Cl, Br, I] reactions using explicitly-correlated ab initio methods providing benchmark stationary-point properties, which may guide future PES developments and dynamics studies [15].

#### Summary

We have studied the dynamics and mechanisms of fundamental chemical reactions based on an efficient analytical PES-based approach, introduced by us for  $S_N2$  reactions. Our major achievements in the past 3 years can be summarized as follows:

- (a) generalization of the double-inversion mechanism,
- (b) uncovering unexpected leaving-group effects,
- (c) high-level ab initio analytical PESs for the  $F^-$  + CH<sub>3</sub>F and  $F^-$  + CH<sub>3</sub>I systems,
- (d) deciphering front-side complex formation in  $S_N 2$  reactions,
- (e) the best quantum dynamics computation for a  $S_N 2$  reaction,
- (f) benchmark ab initio data for several  $S_N 2$  reactions,
- (g) vibrational and rotational mode specificity in  $S_N 2$  reactions,
- (h) mapping the angle-dependent barrier to reaction,
- (i) revealing novel retention pathways for the  $Cl + CH_4$  substitution process.

The above work resulted in 15 research articles (cumulative impact factor: 108, corresponding-author papers: 14) including high-impact publications in *Nature Chemistry* (twice), *Chemical Science*, etc. and an invited Feature Article as well as several conference presentations including 6 invited talks and a Doctor of the Hungarian Academy of Sciences dissertation. Since this is a PD grant, it is worth emphasizing that none of the above-mentioned publications involved my PhD and postdoctoral advisers.

## **References (all supported by the present PD-111900 grant)**

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Benchmark ab initio characterization of the complex potential energy surface of the  $F^- + CH_3 CH_2 Cl$  reaction

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[10] I. Szabó, B. Olasz, and **G. Czakó**\* Deciphering front-side complex formation in  $S_N^2$  reactions via dynamics mapping *J. Phys. Chem. Lett.* 8, 2917 (2017)

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Benchmark ab initio characterization of the complex potential energy surface of the  $\rm Cl^- + CH_3I$  reaction

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Dynamics and novel mechanisms of S<sub>N</sub>2 reactions on ab initio analytical potential energy surfaces
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This paper was featured on the front cover of JPCA

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Benchmark ab initio characterization of the complex potential energy surfaces of the  $X^- + NH_2Y$  [X,Y = F, Cl, Br, I] reactions J. Phys. Chem. A DOI: 10.1021/acs.jpca.7b11927 (2018)

# **Conference presentations**

# <u>G. Czakó</u>

Kémiai reakciók dinamikája ab initio potenciális energia felületeken (in Hungarian) *Meeting of the Material and Molecular Structure Working Group of MTA* Mátrafüred, Hungary, February 27-28 2015 (**Talk**)

## <u>G. Czakó</u>

Dynamics of hydrogen-abstraction and  $S_N 2$  reactions on ab initio analytical potential energy surfaces *XIII International Workshop on Quantum Reactive Scattering* Salamanca, Spain, July 6-10 2015 (Invited talk)

P. Szabó, G. Lendvay, and G. Czakó

Quasiclassical trajectory studies of the dynamics of a complex-forming bimolecular reaction:  $CH_3 + HBr \rightarrow CH_4 + Br$ 

XIII International Workshop on Quantum Reactive Scattering Salamanca, Spain, July 6-10 2015 (Talk)

## G. Czakó

Dynamics of hydrogen-abstraction and  $S_N^2$  reactions on ab initio analytical potential energy surfaces *XXV Dynamics of Molecular Collisions Meeting* Asilomar, CA, USA, July 12-17 2015 (**Poster**)

<u>M. Stei</u>, E. Carrascosa, M. Kainz, A. Kelkar, I. Szabó, **G. Czakó**, A. Dörfler, T. Michaelsen, J. Meyer, and R. Wester Imaging the influence of the leaving group and of vibrational excitation on S<sub>N</sub>2 reactions *XXV Dynamics of Molecular Collisions Meeting* Asilomar, CA, USA, July 12-17 2015 (**Talk**)

# <u>G. Czakó</u>

Dynamics of chemical reactions on ab initio potential energy surfaces 14th Central European Symposium on Theoretical Chemistry (CESTC) Banská Bystrica, Slovakia, September 6-9 2015 (Invited talk)

M. Stei, E. Carrascosa, <u>J. Meyer</u>, T. Michaelsen, B. Bastian, M. A. Kainz, A. H. Kelkar, I. Szabó, G. Czakó, and R. Wester

Imaging the complex influence of the leaving group on nucleophilic substitution reactions *12th European Conference on Atoms Molecules and Photons (ECAMP12)* Frankfurt, Germany, September 5-9 2016 (**Poster**)

M. Stei, E. Carrascosa, J. Meyer, T. Michaelsen, B. Bastian, M. A. Kainz, A. H. Kelkar, I. Szabó, G. Czakó, and R. Wester

Imaging the complex influence of the leaving group on nucleophilic substitution reactions 21st European Conference on the Dynamics of Molecular Systems (MOLEC) Toledo, Spain, September 11-16 2016 (**Poster**)

## <u>G. Czakó</u>

Kémiai reakciók dinamikája ab initio potenciális energia felületeken (in Hungarian)

Meeting of the Reaction Kinetics and Photochemistry Working Group of MTA Balatonalmádi, Hungary, May 29-30 2017 (**Talk**)

## G. Czakó

Dynamics and novel mechanisms of  $S_N 2$  reactions on ab initio analytical potential energy surfaces 8th International Meeting on Atomic and Molecular Physics and Chemistry (IMAMPC) Torun, Poland, June 19-22 2017 (Invited talk)

## <u>G. Czakó</u>

Dynamics and novel mechanisms of  $S_N 2$  reactions on ab initio analytical potential energy surfaces *XIV International Workshop on Quantum Reactive Scattering* Trieste, Italy, July 3-6 2017 (**Invited talk**)

## G. Czakó

Dynamics and novel mechanisms of  $S_N 2$  reactions on ab initio analytical potential energy surfaces  $S_N 2$  Reaction Dynamics Symposium, ACS Southwest Regional Meeting Lubbock, TX, USA, October 29 - November 1 2017 (Invited talk)

## G. Czakó

Dynamics and novel mechanisms of reactions of atoms and ions with polyatomic molecules *WG1 Workshop*, *COST Action CM1401* Ciudad Real, Spain, December 11-12 2017 (**Invited talk**)